

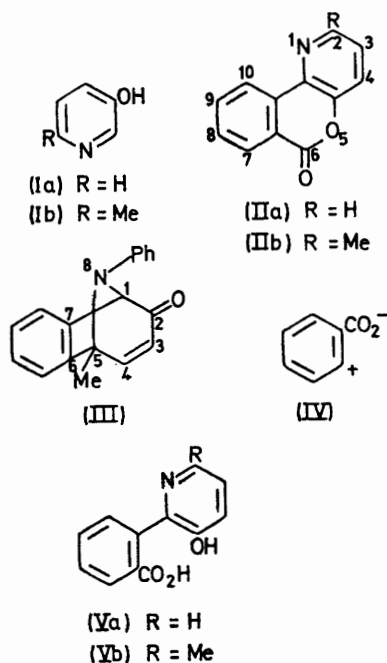
A New Synthesis of Azabenzocoumarins

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Summary 3-Hydroxypyridine (Ia) and its 6-methyl analogue (Ib) react with the benzyne precursor (IV) to give 1-azabenzocoumarins (IIa,b); the latter reaction also yielded 6,7-benzo-5-methyl-8-phenyl-8-azabicyclo[3,2,1]-oct-3-en-2-one (III) as a by-product.

WE now describe a synthesis of the novel 1-azabenzocoumarin ring system that may be of wide applicability in the preparation of coumarin derivatives (*cf.* ref. 1). 3-Hydroxypyridine (Ia) reacts with the benzyne precursor (IV) to produce 6-oxobenzopyrano[3,2-*b*]pyridine (IIa),



m.p. 134—136°.† Similarly, 6-methyl-3-hydroxypyridine (Ib reacts to form the homologous 2-methyl-6-oxobenzopyrano [3,2-*b*]pyridine (IIb), m.p. 138—139°,† together with the cycloadduct (III), m.p. 128—129°, as a by-product expected by analogy with previous results.²

The n.m.r. spectra of the 1-azabenzisocoumarins supported the structural assignments. For instance, 2-methyl-6-oxobenzopyrano[3,2-*b*]pyridine showed δ (CCl₄)‡ 8.76 (dq, 7-H, $J_{7,8}$ 8.00 Hz), 8.41 (dq, 10-H, $J_{7,10}$ 1.00, $J_{9,10}$ 7.75 Hz), 7.86 (td, 9-H, $J_{8,9}$ 7.50, $J_{7,9}$ 1.50 Hz), 7.70 (td, 8-H, $J_{8,10}$ 1.50 Hz), 7.58 (d, 4-H, $J_{3,4}$ 8.50 Hz), 7.30 (d, 3-H, $J_{3,4}$ 8.50 Hz), and 2.71 (s, C-CH₃). These assignments were supported by spin-spin decoupling and lanthanide shift studies.

Intermediates of the type (Va) and (Vb) are probably involved and dehydrate under the reaction conditions to give the azabenzisocoumarins (IIa,b). Benzyne³ was prepared from anthranilic acid and *n*-pentyl nitrite. The benzyne precursor (IV) (or its equivalent)⁴ has been previously postulated as an intermediate in the formation of benzocoumarin⁵ during the pyrolysis of diphenyliodonium-2-carboxylate⁵ and from potassium 2-halogenobenzoates.⁶ In previously recorded cases,⁷ where benzenediazonium-2-carboxylate was decomposed, there was no indication of the presence of (IV).

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† All compounds give satisfactory analyses for C, H, N and structures were elucidated by i.r., u.v., n.m.r., and mass spectroscopy.

‡ Measured at 100 MHz. Chemical shifts relative to Me₄Si.

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